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Background 1.0.

The Hudson Basin is thought to hold 67.3 million recoverable barrels of oil equivalent, according to the Geological Survey of Canada. The Arctic is becoming more accessible for oil exploration due to the recent decreases in sea ice coverage brought about by climate change. This has also opened up new shipping routes in the region and increased vessel traffic. In light of the increased likelihood of oil pollution resulting from these changes, it is increasingly important to understand the rates and mechanisms of the natural attenuation of oil by microorganisms.

1.1. Marine Oil Snow (MOS)

MOS are oil-entrained flocs composed of small organic and inorganic materials as well as prokaryotic and phytoplanktonic cells. MOS transports hydrocarbons to the ocean floor as well as the microbial communities associated with it. MOS particles harbor a bacterial community that is distinctly different from the free-living community present in the surrounding seawater.

2.0. Objectives

1) Measure oil biodegradation in MOS particles and the water phase in experimental incubations by analyzing changes in the oil composition over time.

2) Describe how seasonal changes in temperature, salinity, and microbial composition affect oil biodegradation within MOS particles and water phases in mesocosm incubations at the Churchill Marine Observatory (CMO).



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Biodegradation potential of estuarine microbial communities from the Churchill Marine Observatory in a marine oil snow microcosm experiment

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Methods and Materials 3.0.

Estuarine water samples were collected in August 2024 from the Ocean Sea Ice Mesocosm facility (OSIM) located at CMO, with water derived from the Churchill River Estuary. The salinity of the water was 5 PSU. The water was filled to half the volume of Pyrex glass bottles and amended with Churchill Diesel oil at an oil-to-water ratio of 0.1% (V/V). The bottles were incubated for 28 days rotating at a speed of 10 rpm to mimic natural marine snow formation. The microcosms were sampled over a 4-week period.



Figure 1 : Roller bottle experiment

Chemical Analysis 3.1.

Oil was extracted from each microcosm three times with hexane and dichloromethane and then dried with sodium sulfate.

Then the sample was evaporated in the Rotovap until a final volume of 1 mL was retained for GC-MS analysis. An Agilent 7010B Triple Quadrupole GC-MS (QqQ-MS) was used to analyze alkanes and PAHs using multiple reaction monitoring.

Table 1 : Diagnostic isomer ratios used as indicators of biodegradation.





Figure 3: Agilent 7010B Triple Quadrupole GC-MS (QqQ-MS). Photo credit: Lisa Oswald





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Figure 2: Evaporation of the extract using Rotovap

Table 2 : Percentage loss of select PAHs

Figure 4: Gas chromatograms of n-alkanes on Day 0 and 4-weeks

No MOS formation was observed during the 28-day incubations. However, the comparison of ratios of C17/ Pristane from Day 0 and Week 4 indicated that biodegradation of alkanes did occur in the water from CMO.

Additionally, the comparison of ratios of 2methylnaphthalene/ 1-methylnaphthalene indicated potential PAH degradation. Percentage losses of PAHs were calculated using $[((A_0/C_0)-(A_S/C_S))/(A_0/C_0)] \times 100$, where A_S and C_S are the concentrations of the target compound and conserved compound (2,4dimethylphenanthrene) in the 4-weeks incubation.

Relative to the high molecular weights, the low molecular weight alkanes degrade over time presumably due to microbial degradation (as shown by the gas chromatograms).

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Screening for biosurfactant-producing bacterial species will be carried out using three methods: 1. Emulsifying index

Additional incubations from autumn and spring estuarine waters will be conducted to determine the seasonal cycle of MOS-forming microbes and their biodegradation potentials.

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Discussion 4.0.

5.0. Future Directions

will extract DNA from the microbial nmunity present in the water and conduct tgun metagenomic analysis using nanopore uencing technology.

- Biosurfactant hemolytic activity
- Blue agar plate method

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